

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11)

EP 0 628 589 B1

(12)

EUROPEAN PATENT SPECIFICATION

(45) Date of publication and mention
of the grant of the patent:
12.03.1997 Bulletin 1997/11

(51) Int Cl.⁶: **C08G 77/08, C08G 77/26**

(21) Application number: **94303881.0**

(22) Date of filing: **27.05.1994**

(54) **Process for the preparation of organopolysiloxanes**

Verfahren zur Herstellung von Organopolysiloxanen

Procédé de préparation d'organopolysiloxanes

(84) Designated Contracting States:
DE FR GB

(30) Priority: **03.06.1993 GB 9311509**

(43) Date of publication of application:
14.12.1994 Bulletin 1994/50

(73) Proprietor: **DOW CORNING LIMITED**
South Glamorgan CF63 2YL Wales (GB)

(72) Inventors:
• **Cray, Steven Edward**
Sully, South Glamorgan, Wales (GB)

• **Chowdhry, Mubarik M.**
Southfields, London SW18 5QT (GB)

(74) Representative:
Vandamme, Luc Johan Roger et al
Dow Corning Limited,
Cardiff Road
Barry, South Glamorgan CF63 2YL, Wales (GB)

(56) References cited:
EP-A- 0 382 365 **EP-A- 0 382 366**
EP-A- 0 503 825

EP 0 628 589 B1

Note: Within nine months from the publication of the mention of the grant of the European patent, any person may give notice to the European Patent Office of opposition to the European patent granted. Notice of opposition shall be filed in a written reasoned statement. It shall not be deemed to have been filed until the opposition fee has been paid. (Art. 99(1) European Patent Convention).

Description

This invention relates to a method for the preparation of organopolysiloxanes having silicon-bonded substituents containing amino groups.

Organosiloxane polymers having silicon-bonded aminoalkyl or polyaminoalkyl groups are known. They are employed *inter alia* as components of polish compositions and in the treatment of textile fibres to impart thereto properties such as softness and shrink resistance.

Polymers which may be used for such applications can be prepared from the appropriate cyclic siloxanes by an equilibration reaction. However, such a route results in the formation of a relatively high proportion of volatile compounds in the product rendering it less than satisfactory for some applications. Amino-containing siloxane polymers may also be obtained by reacting together an organosiloxane having silicon-bonded hydroxyl groups and a silane having silicon-bonded alkoxy groups and aminoalkyl or polyaminoalkyl groups, see for example British Patent GB-A-942 587. It was found, however, that when prepared by such a method the viscosity of the polymer increased during storage. A method of reducing the rate of increase in viscosity of such organosiloxanes is disclosed in GB 2 036 052 B wherein the hydroxylated siloxane and the amino-containing silane are reacted together in a specified range of proportions. Although a significant reduction in viscosity drift can be achieved according to the teaching of GB 2 036 052 B there has existed a need for further improvement in this respect.

E. P. 0 382 365 A2 discloses a process for the preparation of an organosilicon condensation product wherein there are reacted together in the presence of a catalytic substance selected from barium hydroxide and strontium hydroxide, an organosilicon compound having at least one silicon-bonded hydroxyl group and an organo-silicon compound having at least one silicon-bonded -OX group in which X represents certain alkyl or alkoxyalkyl groups. Said EP also discloses that residual silanol groups in the condensation product can be condensed by subsequent contact with a second catalyst which may be a borate or phosphate of sodium or potassium. Although the said process can be employed for the preparation of amino-containing organopolysiloxanes it has been found that where the organosilicon compounds having at least one silicon-bonded OX group are silanes, the process yields by-products which are dimers of the silanes. Quite apart from the fact that the dimer material is an impurity and therefore needs to be removed from the reaction mixture, the organosilicon compound thus consumed cannot react with the organosilicon having at least one hydroxyl group and the yield of the desired condensation product is thereby reduced. Furthermore, a process involving the addition of a catalyst in two steps will involve the risk of error at each stage of addition; the risk of error is reduced if the catalyst is added in one step.

We have now found that by mixing together certain reactive organosilicon compounds and a catalyst comprising both at least one compound selected from strontium and barium hydroxide and at least one compound selected from borates and phosphates of sodium and thereafter reacting the mixture at a temperature of at least 50°C we can produce a condensation product with reduced dimer formation.

According to the present invention there is provided a process for the preparation of organopolysiloxanes having silicon-bonded aminoalkyl or polyaminoalkyl groups which comprises mixing together and thereafter reacting at a temperature of at least 50°C (A) a silanol terminated organopolysiloxane compound, (B) an organosilicon compound having at least one silicon-bonded group -OX wherein X is an alkyl group or an alkoxyalkyl group and having a monovalent group composed of carbon, hydrogen and nitrogen said monovalent group having at least one -NH₂ group therein and being attached to silicon by way of a silicon to carbon linkage, and (C) a catalyst comprising both (i) at least one compound selected from barium hydroxide and strontium hydroxide and (ii) at least one compound selected from borates and phosphates of sodium.

The organopolysiloxane compound (A) may be a polydiorganosiloxane according to the general formula $\text{HOSiR}_2(\text{OSiR}_2)_n\text{OH}$ wherein n is from 4 to 100, and R is an organic substituent selected from alkyl having 2 to 8 carbon atoms, for example, ethyl, propyl, pentyl and hexyl, alkenyl having from 2 to 6 carbon atoms, for example vinyl, allyl and hexenyl and phenyl. Preferably at least 80% of the total silicon-bonded substituents are methyl groups.

The organosilicon compound (B) employed in the process of this invention may be a silane or siloxane compound. A mixture of silanes, siloxanes or both may also be used. Preferred organosilicon compounds (B) are silanes represented by the general formula $\text{CH}_3(\text{XO})_2\text{SiY}$ in which X represents, for example, methyl, ethyl, propyl, butyl, amyl and methoxyethyl and preferably represents methyl or ethyl. The substituent Y is a monovalent group composed of carbon, hydrogen, nitrogen and, optionally, oxygen and is bonded to the silicon atom through a Si-C linkage. Provided at least one -NH₂ group is present in Y any additional nitrogen may be present as, for example -NH- or -N-. Specific examples of Y substituents are $-(\text{CH}_2)_3\text{NH}_2$, $-(\text{CH}_2)_4\text{NH}_2$, $-(\text{CH}_2)_3\text{NH}(\text{CH}_2)_2\text{NH}_2$, $-\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{NH}(\text{CH}_2)_2\text{NH}_2$, $-(\text{CH}_2)_3\text{NHCH}_2\text{CH}_2\text{NH}(\text{CH}_2)_2\text{NH}_2$, $-(\text{CH}_2)_3\text{NH}(\text{CH}_2)_4\text{NH}_2$ and $-(\text{CH}_2)_3\text{O}(\text{CH}_2)_2\text{NH}_2$. Preferred as the Y substituents are those which can be represented by the general formula -R'NHR" in which R' represents an alkylene group having 3 or 4 carbon atoms and R" represents a hydrogen atom or the group $-(\text{CH}_2)_2\text{NH}_2$.

The catalyst (C) comprises in combination (i) at least one compound selected from barium hydroxide and strontium hydroxide together with (ii) at least one compound selected from borates and phosphates of sodium. Specific examples

of (ii) are $\text{Na}_2\text{B}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$, $\text{NaBO}_3 \cdot 4\text{H}_2\text{O}$, $\text{NaBO}_2 \cdot x\text{H}_2\text{O}$ and $\text{Na}_3\text{PO}_4 \cdot 12\text{H}_2\text{O}$. A preferred catalyst (C) is barium hydroxide in combination with trisodium phosphate ($\text{Na}_3\text{PO}_4 \cdot 12\text{H}_2\text{O}$). The relative proportions of (i) and (ii) may be varied within wide limits, for example the ratio of (i) to (ii) may be from 1:9 to 9:1. Sufficient amount of (C) should be employed to provide by weight from 0.01 to 1%, preferably from 0.02 to 0.1%, of each of (i) and (ii) based on the combined weights of (A), (B) and C in the reaction mixture.

In the performance of the process of this invention organopolysiloxane (A), organosilicon compound (B) and catalyst (C) are mixed together and (A) and (B) are allowed to react at a temperature of at least 50°C . The temperature at which the reaction is caused to proceed is not critical but having regard to the volatility of (B) and considerations of energy economy the temperature employed preferably does not exceed about 150°C . The reaction components may be mixed in any convenient order. Generally it is most practical to add the organosilicon compound (B) and the catalyst (C) to the organopolysiloxane (A) which may, if desired be preheated prior to such addition. The catalyst (C) may be incorporated with the other reactants in the form of a mixture of (i) and (ii). However, no advantage is apparent from such premixing and it is preferred to add (i) and (ii) individually and simultaneously to (A) or to a mixture of (A) and (B). Some delay between the incorporation of (i) and (ii) into the reaction mixture can be tolerated provided that the temperature of the mixture has not reached that at which significant reaction between (A) and (B) occurs. Typically the components (i) or (ii) or (i) and (ii) combined are added to the reaction mixture at a temperature of 50°C or below to 95°C , e.g. 85°C .

The process of this invention is best performed under reduced pressure and with removal of water and alcohol by-products during the course of the reaction. When the desired degree of reaction has been achieved volatile substances may be removed from the product under reduced pressure.

In preparing organopolysiloxanes by the process of this invention the relative amounts of organopolysiloxane compound (A) and organosilicon compound (B) employed will depend on the desired content of amine-containing siloxane units in the product and on the relative molecular weights of (A) and (B). For most applications it is preferred that the reaction product contain from about 0.5 to 5 mol percent of such amine-containing units. In general therefore we prefer to carry out the reaction employing from about 1 to 10 parts by weight of organosilicon compound (B) per 100 parts by weight of (A).

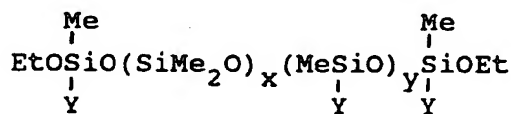
Organopolysiloxanes prepared by the process of this invention find application in the variety of uses known for aminoalkyl- or polyaminoalkyl-siloxanes, for example as ingredients of polishes or for imparting softness or other desirable properties to textile fibres. They are characterised by the presence of reactive silicon-bonded alkoxy, alkoxy-alkoxy or hydroxyl groups and may thus be crosslinked to elastomers by the addition of suitable organosilicone crosslinking agents and, where necessary, catalysts.

The invention is illustrated by the following Example in which the parts are expressed by weight, viscosity is given at 25°C and Me and Et represent methyl and ethyl groups respectively.

Example 1

A polydimethylsiloxane (1) having terminal silanol groups and a viscosity of 72cS, (72mm²/s) (953g) was placed in a reaction flask fitted with a thermometer, stirrer and reflux condenser and heated to 85°C . To the polydimethylsiloxane was then added $(\text{EtO})_2\text{MeSi}(\text{CH}_2)_3\text{NH}_2$ (47g), Na_3PO_4 (0.25g) and $\text{Ba}(\text{OH})_2 \cdot 8\text{H}_2\text{O}$ (0.75g). The stirred reaction mixture was refluxed for 2 hours at 85°C under atmospheric pressure and then under reduced pressure (300 mbar) with removal of volatile by-products until a cooled sample of the product (25°C) had a viscosity of 800cS (800mm²/s).

The reaction mixture was then allowed to cool to 40°C and a vacuum applied to remove remaining volatiles and leave a polydiorganosiloxane having the formula



in which $x = 400$, $y = 10$ and Y represents the group $\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$, and having a viscosity of 2000mm²/s. No dimerised silane was present as determined by ^{29}Si (70.4MHz) nmr.

Comparative Example.

Polydiorganosiloxane as used in Example 1 (953g) was placed in a reaction vessel fitted with a thermometer, stirrer and reflux condenser and was heated to a temperature of 85°C . $(\text{EtO})_2\text{MeSi}(\text{CH}_2)_3\text{NH}_2$ (47g) and $\text{Ba}(\text{OH})_2 \cdot 8\text{H}_2\text{O}$ (0.75g) were added to the polydiorganosiloxane (1) and the resultant mixture was stirred, at 85°C for 2 hours

at atmospheric pressure. Na_3PO_4 (0.25g) was then added and, under reduced pressure (300 mbar) volatile by-products were removed until a cooled sample of the product (25°C) showed a viscosity of 150mm²/s.

The reaction mixture was then allowed to cool to 40°C and a vacuum applied to remove remaining volatiles.

The resultant polymer had a viscosity of 2000mm²s⁻¹. Dimerised silane was found to be present at a level of 0.1mol%

Claims

1. A process for the preparation of organopolysiloxanes having silicon-bonded aminoalkyl or polyaminoalkyl groups which comprises mixing together and thereafter reacting at a temperature of at least 50°C (A) a silanol end-blocked organopolysiloxane, (B) an organosilicon compound having at least one silicon-bonded group -OX wherein X is an alkyl group or an alkoxyalkyl group and having a monovalent group composed of carbon, hydrogen and nitrogen said monovalent group having at least one -NH₂ group therein and being attached to silicon by way of a silicon to carbon linkage, and (C) a catalyst characterised in that the catalyst comprises both (i) at least one compound selected from barium hydroxide and strontium hydroxide and (ii) at least one compound selected from borates and phosphates of sodium.
2. A process according to Claim 1 further characterised in that the catalyst (C) consists of barium hydroxide in combination with trisodium phosphate.
3. A process according to either of the Claims 1 and 2 further characterised in that (i) and (ii) are each employed in an amount of from 0.01 to 1.0% by weight based on the combined weights of (A), (B) and (C).
4. A process according to any one of the preceding claims further characterised in that (A), (B) and (C) are reacted at a temperature of from 50 to 150°C.
5. A process according to any one of the preceding claims further characterised in that the organopolysiloxane (A) is a polydiorganosiloxane according to the general formula $\text{HOSiR}_2(\text{OSiR}_2)_n\text{OH}$ wherein n has a value from 4 to 100 and R is an alkyl group having from 2 to 8 carbon atoms, an alkenyl group having from 2 to 6 carbon atoms or a phenyl group.
6. A process according to Claim 5 further characterised in that at least 80% of the R groups in organopolysiloxane (A) are methyl groups.
7. A process according to any one of the preceding Claims further characterised in that the organosilicon compound (B) is a silane according to the formula $\text{CH}_3(\text{OX})_2\text{SiY}$ wherein each X represents an alkyl group or an alkoxyalkyl group and Y represents a monovalent group composed of carbon, hydrogen and nitrogen, said Y group having at least one -NH₂ group therein and being attached to silicon by way of a silicon-to-carbon linkage.

Patentansprüche

1. Verfahren zur Herstellung von Organopolysiloxanen mit siliciumgebundenen Aminoalkyl- oder Polyaminoalkylgruppen, das umfaßt, daß man
 - (A) ein Organopolysiloxan mit Silanolendgruppen,
 - (B) eine Organosiliciumverbindung mit mindestens einer siliciumgebundenen Gruppe -OX, worin X eine Alkylgruppe oder eine Alkoxyalkylgruppe ist, die eine monovalente Gruppe aus Kohlenstoff, Wasserstoff und Stickstoff aufweist, wobei die monovalente Gruppe mindestens eine -NH₂-Gruppe aufweist und an Silicium durch eine Silicium-Kohlenstoff-Bindung gebunden ist, und
 - (C) einen Katalysator miteinander vermischt und danach bei einer Temperatur von mindestens 50°C umsetzt, dadurch gekennzeichnet, daß der Katalysator sowohl
 - (i) mindestens eine Verbindung ausgewählt aus Bariumhydroxid und Strontiumhydroxid und
 - (ii) mindestens eine Verbindung ausgewählt aus Boraten und Phosphaten von Natrium umfaßt.
2. Verfahren nach Anspruch 1, weiter dadurch gekennzeichnet, daß der Katalysator (C) aus Bariumhydroxid und

Trinatriumphosphat besteht.

3. Verfahren nach einem der Ansprüche 1 oder 2, weiter dadurch gekennzeichnet, daß (i) und (ii) jeweils in einer Menge von 0,01 bis 1,0 Gew.-%, bezogen auf das vereinigte Gewicht von (A), (B) und (C) angewendet werden.
4. Verfahren nach einem der vorhergehenden Ansprüche, weiter dadurch gekennzeichnet, daß (A), (B) und (C) bei einer Temperatur von 50 bis 150°C umgesetzt werden.
5. Verfahren nach einem der vorhergehenden Ansprüche, weiter dadurch gekennzeichnet, daß das Organopolysiloxan (A) ein Polydiorganosiloxan der allgemeinen Formel $\text{HOSiR}_2(\text{OSiR}_2)_n\text{OH}$ ist, worin n einen Wert von 4 bis 100 hat und R eine Alkylgruppe mit 2 bis 8 Kohlenstoffatomen, eine Alkenylgruppe mit 2 bis 6 Kohlenstoffatomen oder eine Phenylgruppe ist.
6. Verfahren nach Anspruch 5, weiter dadurch gekennzeichnet, daß mindestens 80% der Gruppen R in dem Organopolysiloxan (A) Methylgruppen sind.
7. Verfahren nach einem der vorhergehenden Ansprüche, weiter dadurch gekennzeichnet, daß die Organosiliciumverbindung (B) ein Silan der Formel $\text{CH}_3(\text{OX})_2\text{SiY}$ ist, worin jeder Rest X eine Alkylgruppe oder eine Alkoxyalkylgruppe bedeutet und Y eine monovalente Gruppe aus Kohlenstoff, Wasserstoff und Stickstoff bedeutet, wobei die Gruppe Y mindestens eine $-\text{NH}_2$ -Gruppe enthält, die an Silicium über eine Silicium-Kohlenstoff-Bindung gebunden ist.

Revendications

1. Un procédé de préparation d'organopolysiloxanes ayant des groupes aminoalkyles ou polyaminoalkyles liés au silicium selon lequel on mélange ensemble et ensuite on fait réagir à une température d'au moins 50°C (A) un organopolysiloxane bloqué terminalement par des groupements silanols, (B) un composé organique du silicium ayant au moins un groupe $-\text{OX}$ lié au silicium dans lequel X est un groupe alkyle ou un groupe alcoxyalkyle et ayant un groupe monovalent composé de carbone, d'hydrogène et d'azote, ledit groupe monovalent comportant au moins un groupe $-\text{NH}_2$ et étant relié au silicium au moyen d'une liaison silicium-carbone, et (C) un catalyseur caractérisé en ce que le catalyseur comporte à la fois (i) au moins un composé choisi parmi l'hydroxyde de baryum et l'hydroxyde de strontium et (ii) au moins un composé choisi parmi les borates et les phosphates de sodium.
2. Un procédé selon la revendication 1 caractérisé de plus en ce que le catalyseur (C) est constitué d'hydroxyde de baryum combiné à du phosphate trisodique.
3. Un procédé selon l'une ou l'autre des revendications 1 et 2 caractérisé de plus en ce que (i) et (ii) sont chacun employés en une quantité comprise entre 0,01 et 1,0% en poids basée sur les poids combinés de (A), (B) et (C).
4. Un procédé selon l'une quelconque des revendications précédentes caractérisé de plus en ce que l'on fait réagir (A), (B) et (C) à une température comprise entre 50 et 150°C.
5. Un procédé selon l'une quelconque des revendications précédentes caractérisé de plus en ce que l'organopolysiloxane (A) est un polydiorganosiloxane de formule générale $\text{HOSiR}_2(\text{OSiR}_2)_n\text{OH}$ dans laquelle n vaut 4 à 100 et R est un groupe alkyle ayant 2 à 8 atomes de carbone, un groupe alcényle ayant 2 à 6 atomes de carbone ou un groupe phényle.
6. Un procédé selon la revendication 5 caractérisé de plus en ce qu'au moins 80% des groupes R dans l'organopolysiloxane (A) sont des groupes méthyles.
7. Un procédé selon l'une quelconque des revendications précédentes caractérisé de plus en ce que le composé organique du silicium (B) est un silane de formule $\text{CH}_3(\text{OX})_2\text{SiY}$ dans laquelle chaque X représente un groupe alkyle ou un groupe alcoxyalkyle et Y représente un groupe monovalent composé de carbone, d'hydrogène et d'azote, ledit groupe Y comportant au moins un groupe $-\text{NH}_2$ et étant relié au silicium au moyen d'une liaison silicium-carbone.

THIS PAGE BLANK (USPTO)